

**Study Material: 1**

Name- Sayanwita Panja

Discipline- B.Sc (H)

Subject- Chemistry

Semester- IV

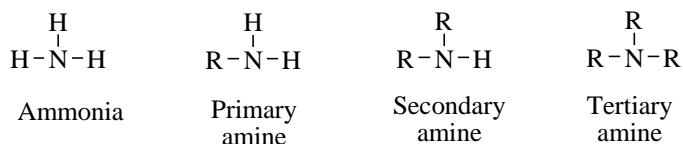
Course Code- CC10

Topic- Organic Chemistry, Nitrogen Compounds

## ORGANIC COMPOUNDS OF NITROGENS:

### Aliphatic Amines:

Amines may be regarded as derivatives of ammonia, in which one or more hydrogen atoms have been replaced by alkyl groups.

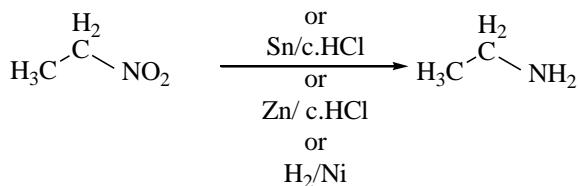


- Isomerism:

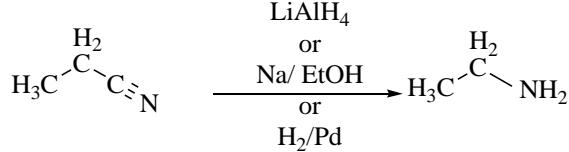
Chain Isomerism	Position Isomerism	Functional Isomerism	Metamerism
$\begin{array}{c} \text{H}_2 \\   \\ \text{H}_3\text{C}-\text{C}-\text{C}-\text{NH}_2 \\   \quad   \\ \text{H}_2 \quad \text{H}_2 \end{array}$ n-Butylamine	$\begin{array}{c} \text{H}_2 \\   \\ \text{H}_3\text{C}-\text{C}-\text{C}-\text{NH}_2 \\   \quad   \\ \text{H}_2 \quad \text{H}_2 \end{array}$ 1-Aminopropane	$\text{CH}_3.\text{CH}_2.\text{CH}_2.\text{NH}_2$ Propylamine (Primary)	$\begin{array}{c} \text{C}_2\text{H}_5-\text{N}-\text{C}_2\text{H}_5 \\   \\ \text{H} \end{array}$ Diethylamine
$\begin{array}{c} \text{H}_2 \\   \\ \text{H}_3\text{C}-\text{CH}-\text{C}-\text{NH}_2 \\   \quad   \\ \text{CH}_3 \quad \text{NH}_2 \end{array}$ Isobutylamine	$\begin{array}{c} \text{H}_2 \\   \\ \text{H}_3\text{C}-\text{CH}-\text{CH} \\   \quad   \\ \text{NH}_2 \quad \text{CH}_3 \end{array}$ 2-Aminopropane	$\text{CH}_3.\text{NH}.\text{C}_2\text{H}_5$ Methylethylamine (Secondary) $(\text{CH}_3)_3\text{N}$ Triethylamine (Tertiary)	$\begin{array}{c} \text{C}_3\text{H}_7-\text{N}-\text{CH}_3 \\   \\ \text{H} \end{array}$ Methylpropylamine

- General methods of preparation of amines:

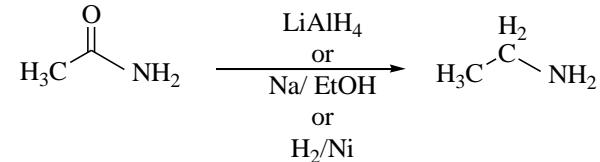
1.



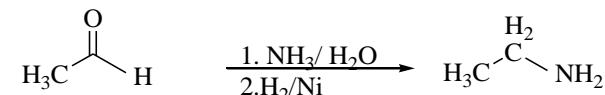
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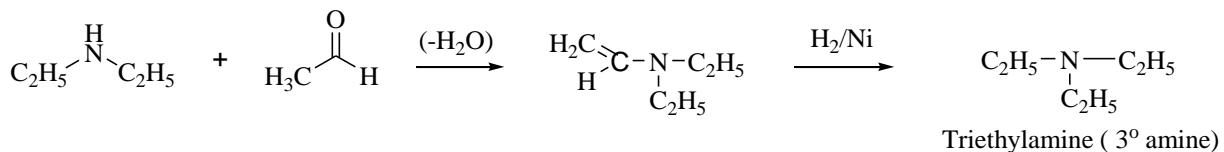
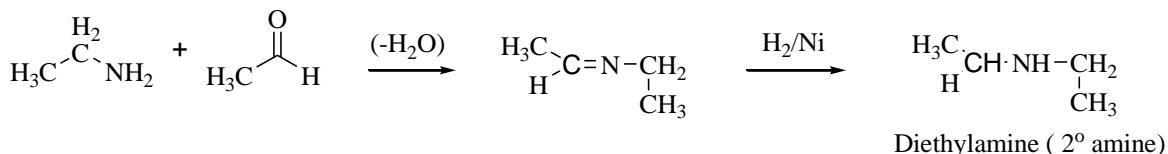
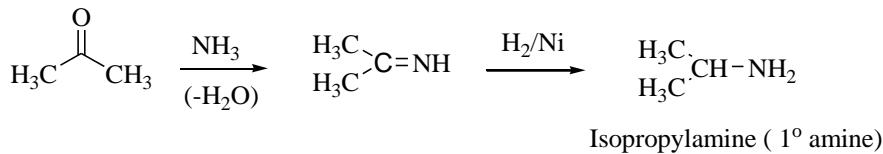
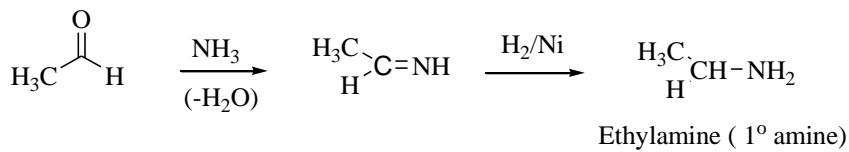
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4.



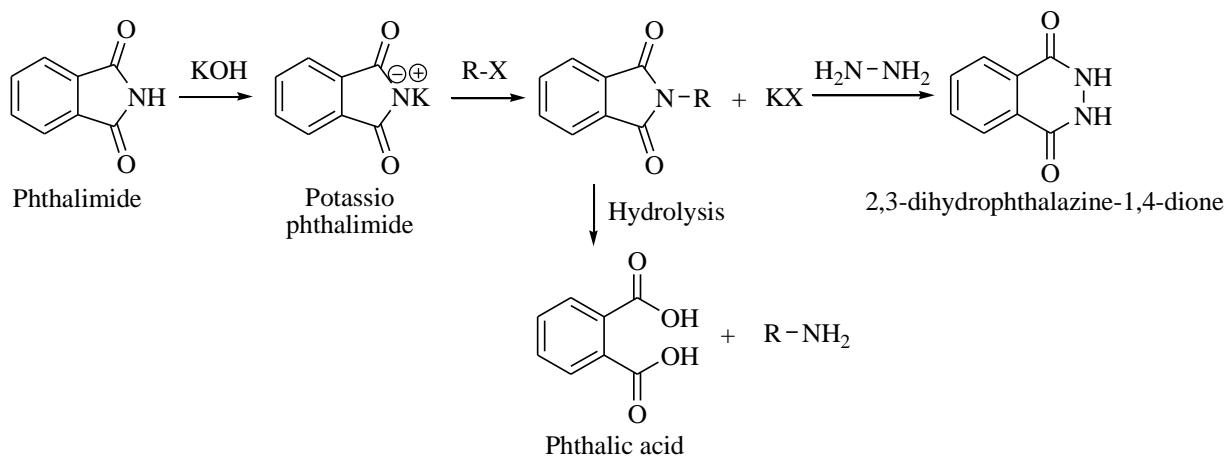
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### 5. Reaction of Chloramine on Grignard reagent:



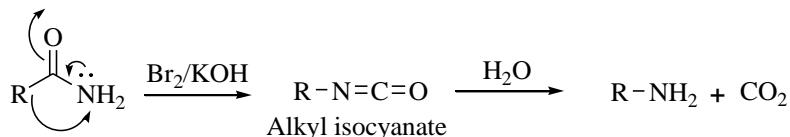
### 6. Gabriel's Phthalimide Synthesis: (1° amine synthesis)



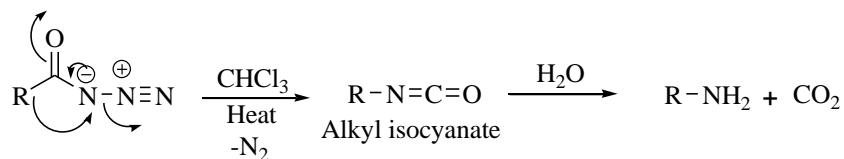
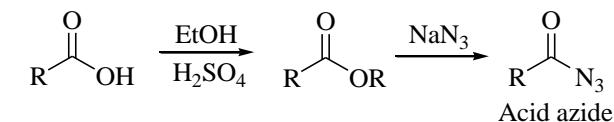
**7. Hofmann, Curtius, Lossen, Schmidt Reaction: (1° amine synthesis)**

Rearrangement	Substrate	Reagent	Intermediate	Product
Hofmann	$\text{R}-\text{C}(=\text{O})-\text{NH}_2$	$\text{Br}_2/\text{KOH}$	$\text{R}-\text{N}=\text{C}=\text{O}$	$\text{R}-\text{NH}_2$
Curtius	$\text{R}-\text{C}(=\text{O})-\text{N}_3$	Heating in $\text{CHCl}_3$	$\text{R}-\text{N}=\text{C}=\text{O}$	$\text{R}-\text{NH}_2$
Lossen	$\text{R}-\text{C}(=\text{O})-\text{NH}-\text{O}-\text{C}(=\text{O})-\text{R}$	$\text{C}_2\text{H}_5\text{ONa}$	$\text{R}-\text{N}=\text{C}=\text{O}$	$\text{R}-\text{NH}_2$
Schmidt	$\text{R}-\text{C}(=\text{O})-\text{OH} + \text{N}_3\text{H}$	$\text{H}_2\text{SO}_4$	$\text{R}-\text{N}=\text{C}=\text{O}$	$\text{R}-\text{NH}_2$

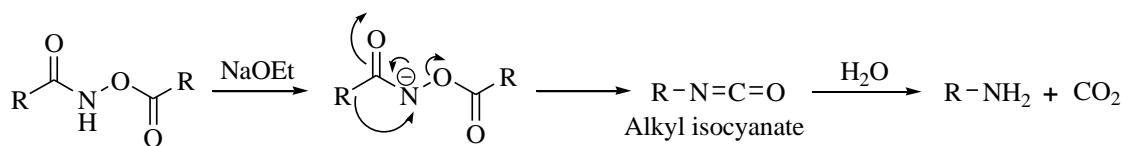
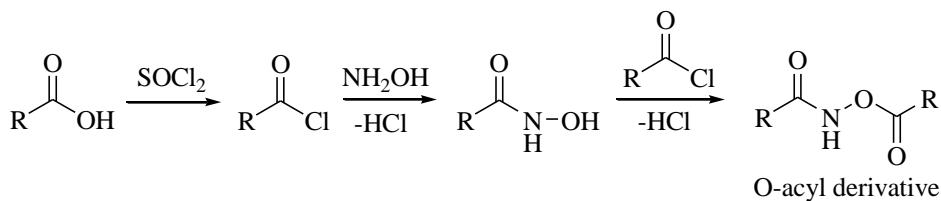
**i. Hofmann Reaction:**



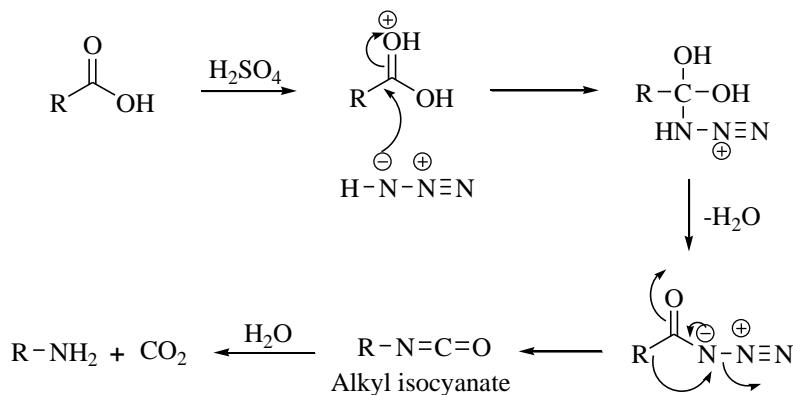
**ii. Curtius Reaction:**



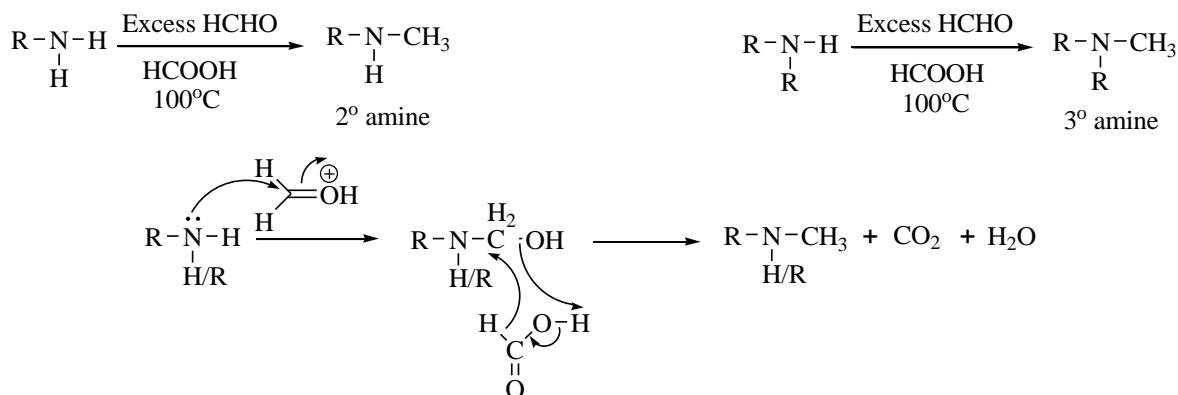
**iii. Lossen Reaction:**



**iv. Schmidt Reaction:**

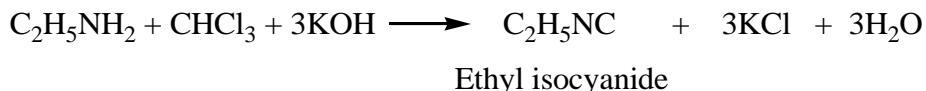


**8. Eschwieler Clarke Methylation: (E-Clarke Reaction,  $2^\circ$  or  $3^\circ$  amine synthesis)**

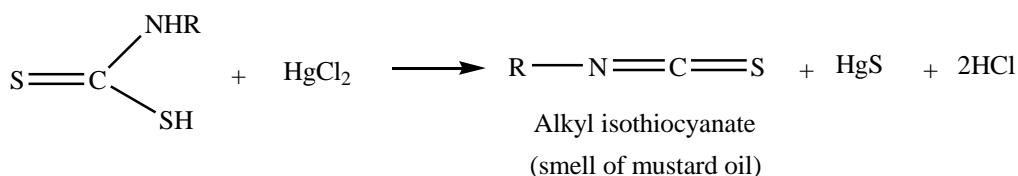
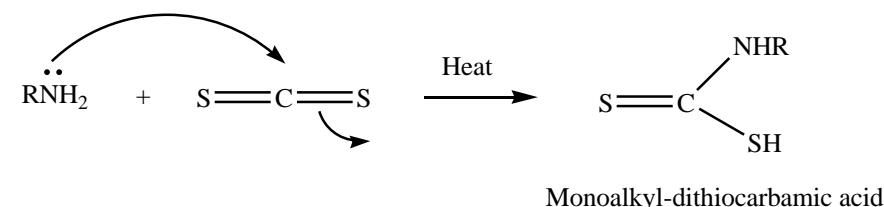


- Reaction of Primary Amine:**

- Carbylamine Reaction:**



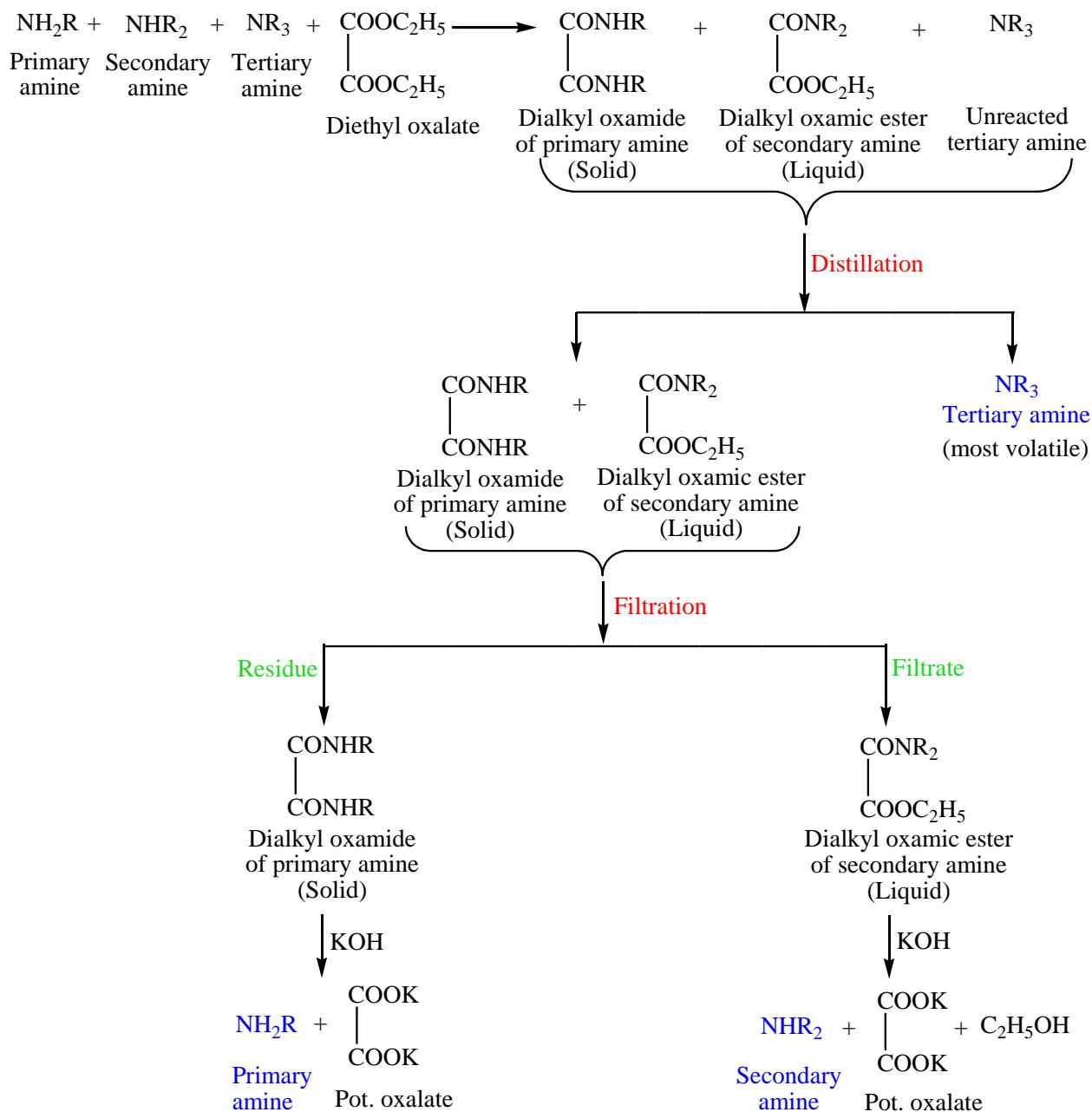
- Hofmann's Mustard Oil Reaction:**



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- **Separation of mixture of Amines:**

1. **Hofmann Method:**



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### 2. Hinsberg Method:

